

Concept Review Section How Are Elements Organized Answers



Concept Review Section How Are

Concept Review j j. Section: What is Chemistry Complete each statement below by underlining the correct word or phrase in brackets. 1. A chemical is any substance that has [definite, indefinite] composition. Changes in chemicals, or chemical reactions, take place [only in test tubes, all]. 2.

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Concept Review: Covalent Bonds. 1. A covalent bond forms when two or more valence electrons are attracted by the positively charged nuclei of two atoms and thus are shared between both atoms. 2. The H₂ molecule is stable because each hydrogen atom now has a shared pair of electrons and has achieved a stable noble gas configuration.

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SECTION: REACTION TYPES 1. oxygen 2. A single-replacement reaction will have one compound and one element as reactants and one compound and one element as products, while a double-replacement reaction has two compounds as reactants and two compounds as products. 3. Electrons are transferred between elements during the reaction. The sub-

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Holt Chemistry 5 The Mole and Chemical Composition. Section: Formulas and Percentage Composition. Solve the following problems, and write your answer in the space provided. 1. The mineral greenockite is a yellow sulfide of cadmium that is 78.0% cadmium and 22.0% sulfur. Determine the empirical formula of greenockite.

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CONCEPT REVIEW SECTION HOW SUBSTANCES DISSOLVE ANSWERS

Concept Review: Studying Matter and Energy. 1. observing, formulating hypotheses, testing hypotheses, analyzing results, drawing conclusions, publishing results 2. The research findings of any experiment or investigation must be reproducible by other scientists for those findings to be accepted as valid.

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Section: Substances Are Made of Atoms. In the blank at the left of each word or phrase, write the letter of the expression on the right that is most closely related. _____ 1. atomic theory a. This states that a chemical compound always contains the same elements in exactly the same proportions by weight or mass.

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Concept Review continued 6. Why was Moseley able to resolve the discrepancies in Mendeleev's table when Mendeleev could not? 7. Explain the importance of valence electrons. 8. Why do elements with similar properties appear at regular intervals in the periodic table? 9. How is the electron configuration similar for each element in a group? 10.

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Section: Limiting Reactants and Percentage Yield Complete each statement below by choosing a term from the following list. ... Concept Review: Stoichiometry and Cars 1. c 2. b 3. d 4. b 5. d 6. If there is too much oxygen and not enough gasoline, the engine will stall.

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Concept Review continued 6. Determine the mass in grams of 7.20 mol of antimony. 7. Determine the mass in grams of 0.500 mol of uranium. 8. Determine the mass in grams of 0.750 mol of francium. 9. A sample of lead has a mass of 150.0 g. What amount of lead in moles does the sample contain? 10. A sample of gold has a mass of 5.00 10³ g. What ...

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Concept Review: Trends in the Periodic Table 1. ionization energy 2. bond radius 3. electron affinity 4. electronegativity 5. increases, decreasing 6. d 7. a 8. c 9. The electron cloud model is based on the probability of finding an electron at a specific location. As you move far-ther out from the nucleus, the proba-bility of finding electrons ...

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Concept Review continued 3. Table 1 shows that [H³⁰] and pH are related. As [H³⁰] decreases from 10⁻⁸ M to 10⁻¹⁰ M, pH from 8 to 10. In general, as the concentration decreases, the increases. In other words, the the pH, the less acidic (more basic) the solution. 4. A solution of pH 6 has a(n) [H³⁰] than does a solution of pH 4.

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Concept Review: Studying Matter and Energy. 1. observing, formulating hypotheses, testing hypotheses, analyzing results, drawing conclusions, publishing results 2. The research findings of any experi- ment or investigation must be repro- ducible by other scientists for those findings to be accepted as valid.

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Concept Review Section: Circuits 1. Identify the types of elements in the schematic diagram shown below. Give the number of each type of element in the diagram. 2. Predict which fuse would give the greatest protection against high current in a circuit—a 20 A fuse or a 30 A fuse. Explain your answer.

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SECTION: NEWTON'S THIRD LAW 1. equal; The force ont he tube is equal and opposite the force on the shell. 2. equal; The force on the skater is equal and opposite the force on the ball. The ball has a smaller mass than the skater and so accelerates more under the same-sized force. Concept Review SECTION: LAWS OF MOTION 1. a. Yes b. c. No d. Yes

Concept Review - Hays High Indians

Concept Review: Describing Chemical Reactions 1. A chemical reaction is the process by which one or more substances change to produce one or more different sub-stances. 2. Answers will vary but

may include iron rusting, milk souring, or wood burning. 3. a. release of energy as heat b. release of energy as light c. production of sound

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